Chapter 3.1: Heat

Body Temperature

- The average body temperature is 37°C.
- If the body gets a few degress warmer or colder, we die.
- The body has a way of keeping constant body temperature.
- Example:
 - Sweating
 - Shivering
- Clothes slow down the transfer of heat energy between our bodies and the surroundings.

Hot Stuff

- Heat and temperature are related, BUT they are not the same thing.
- Temperature: Describes how hot or cold an object is and is usually measured in degrees Celsius (°C).
- Heat: A form of energy and is measured in Joules (J). It is also known as thermal energy.

The amount of heat energy stored in something depends on: Its temperature Its material Its mass

Hot Stuff

- Heat energy always flows from a hot object to a cool one.
 - The larger the temperature difference, the faster the rate of heat transfer.
- The cool object becomes hotter and the hot object becomes cooler until they are both at the same temperature.

Heating Solids

- A conductor is something that lets energy flow through it.
- When something allows heat energy to flow through it, it is called a conductor.
- Some materials are better conductors of heat than others, such as metals.
- A bad conductor of heat is called an insulator.
 - Example: Wood, Plastic

Example

- Do your feet feel colder on tiles or on a carpet?
- Tiles feel colder as they are better conductors of heat than carpets.
- Heat is conducted away from your feet to the tiles.

Question

- If a cup of coffee and a red popsickle were left on the table in this room what would happen to them? Why?
- The cup of coffee will cool until it reaches room temperature. The popsickle will melt and then the liquid will warm to room temperature.

Heat Transfer Methods

Heat transfers in three ways:
Conduction
Convection
Radiation

Conduction

When you heat a metal strip at one end, the heat travels to the other end.



As you heat the metal, the particles vibrate, these vibrations make the adjacent particles vibrate, and so on and so on, the vibrations are passed along the metal and so is the heat. We call this? Conduction

The outer e<u>lectrons</u> of metal atoms drift, and are free to move.



When the metal is heated, this 'sea of electrons' gain k<u>inetic</u> energy and transfer it throughout the metal.



Insulators, such as w<u>ood</u> and p<u>lastic</u>, do not have this 'sea of electrons' which is why they do not conduct heat as well as metals.

Why does metal feel colder than wood, if they are both at the same temperature?

Metal is a conductor, wood is an insulator. Metal conducts the heat away from your hands. Wood does not conduct the heat away from your hands as well as the metal, so the wood feels warmer than the metal.



Convection

What happens to the particles in a liquid or a gas when you heat them?

The particles spread out and become less dense.



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Fluid movement



Cooler, more d<u>ense</u>, fluids sink through w<u>armer</u>, less dense fluids.

In effect, warmer liquids and gases rise_up.

Cooler liquids and gases s<u>ink</u>.

Water movement





Cooler water sinks







Why is it windy at the seaside?



Where is the freezer compartment put in a fridge? It is put at the top, because cool air sinks, so it cools the food on the way down.



Freezer compartment It is warmer at the bottom, so this warmer air rises and a convection current is set up.

The third method of heat transfer

How does heat energy get from the Sun to the Earth?



There are no particles between the Sun and the Earth so it CANNOT travel by conduction or by convection.

Radiation travels in straight lines True/False Radiation can travel through a vacuum True/False Radiation requires particles to travel True/False

Radiation travels at the speed of light True/False

Emission experiment

Four containers were filled with warm water. Which container would have the warmest water after ten minutes?



The <u>shiny metal</u> container would be the warmest after ten minutes because its shiny surface reflects heat <u>indiation</u> back into the container so less is lost. The <u>dull black</u> container would be the coolest because it is the best at <u>omitting</u> heat radiation.

Absorption experiment

Four containers were placed equidistant from a heater. Which container would have the warmest water after ten minutes?



The <u>dull black</u> container would be the warmest after ten minutes because its surface absorbs heat <u>radiation</u> the best. The <u>shiny metal</u> container would be the coolest because it is the poorest at <u>absorbing</u> heat radiation. Why does hot air rise and cold air sink?

Cool air is more dense than warm air, so the cool air 'falls through' the warm air.

Why are boilers placed beneath hot water tanks in people's homes?

Hot water rises.

So when the boiler heats the water, and the hot water rises, the water tank is filled with hot water.

Why are houses painted white in hot countries?

White reflects heat radiation and keeps the house cooler.

Why are shiny foil blankets wrapped around marathon runners at the end of a race?

The shiny metal reflects the heat radiation from the runner back in, this stops the runner getting cold.

1. Which of the following is not a method of heat transfer?

A.Radiation B.Insulation C.Conduction D.Convection 2. In which of the following are the particles closest together?

A.Solid B.Liquid C.Gas D.Fluid

3. How does heat energy reach the Earth from the Sun?

A.Radiation B.Conduction C.Convection D.Insulation 4. Which is the best surface for reflecting heat radiation?

A.Shiny white B. Dull white C. Shiny black D.Dull black 5. Which is the best surface for absorbing heat radiation?

A.Shiny white B.Dull white C.Shiny black D.Dull black